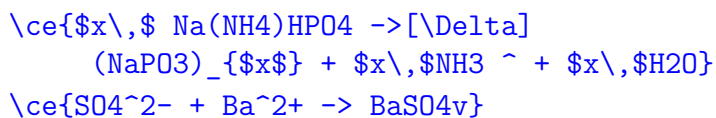
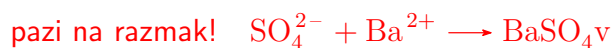
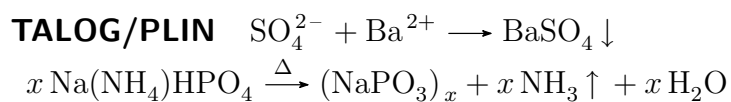
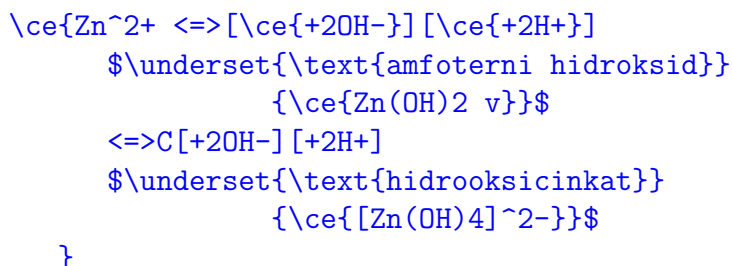
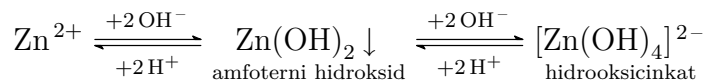


Osnovne naredbe paketa mhchem

OSNOVNO	$\frac{1}{2}\text{H}_2\text{O}$	<code>\ce{1/2H2O}</code>
	$3\text{Cr}_2\text{O}_7^{2-}$	<code>\ce{3Cr2O7^2-}</code>
može i ovako	$3\text{Cr}_2\text{O}_7^{2-}$	<code>\ce{3 Cr2 O7^2-}</code>
ali	$3\text{Cr}_2\text{O}_{7_2}^-$	<code>\ce{3Cr 2O7^ 2-}</code>
treba biti	$3\text{Cr}_2\text{O}_{7_2}^-$	<code>\ce{3Cr 2O7 2-}</code>
pažljiv	$3\text{Cr}_2\text{O}_7^{2-}$	<code>\ce{3Cr2O 7^2-}</code>
s razmacima	$3\text{Cr}_2\text{O}_7^{2-}$	<code>\ce{3 Cr2 O7^2 -}</code>
	$(\text{NH}_4)_2\text{S}$	<code>\ce{(NH4)2S}</code>
IZOTOPI	${}^{227}_{90}\text{Th}^+$	<code>\ce{^{227}_{90}Th+}</code>
FONTOVI	H_2O	<code>\ce{H2O}</code>
	H_2O	<code>\sffamily \ce{H2O}</code>
	H_2O	<code>\sffamily \$\ce{H2O}\$</code>
VEZE	$\text{C}_6\text{H}_5\text{-CHO}$	<code>\ce{C6H5-CHO}</code>
	$\text{A-B=C}\equiv\text{D}$	<code>\ce{A-B=C#D}</code>
	$\text{A-B=C}\equiv\text{D}$	<code>\ce{A\sbond B\dbond C\tbond D}</code>
	$\text{A-B=C}\equiv\text{D}$	<code>\ce{A\bond{-}B\bond{=}C\bond{\#}D}</code>
	$\text{A}\cdots\text{B}\cdots\text{C}\equiv\text{D}$	<code>\ce{A\bond{~}B\bond{~-}C\bond{~-}D}</code>
	$\text{A}\equiv\text{B}\equiv\text{C}\equiv\text{D}$	<code>\ce{A\bond{~=}B\bond{~-}C\bond{~-}D}</code>
	$\text{A}\cdots\text{B}\cdots\text{C}$	<code>\ce{A\bond{...}B\bond{...}C}</code>
	$\text{A}\rightarrow\text{B}\leftarrow\text{C}$	<code>\ce{A\bond{->}B\bond{<-}C}</code>
REAKCIJE	$\text{CO}_2 + \text{C} \rightarrow 2\text{CO}$	<code>\ce{CO2 + C -> 2CO}</code>
a ne	$\text{CO}_2^+\text{C}^- \rightarrow 2\text{CO}$	<code>\ce{CO2+C->2CO}</code>
ili čak	$\text{CO}_2^+\text{C}^- > 2\text{CO}$	<code>\ce{CO2+C- >2CO}</code>
	$\text{CO}_2 + \text{C} \leftarrow 2\text{CO}$	<code>\ce{CO2 + C <- 2CO}</code>
	$\text{CO}_2 + \text{C} \rightleftharpoons 2\text{CO}$	<code>\ce{CO2 + C <=> 2CO}</code>
	$\text{H}^+ + \text{OH}^- \rightleftharpoons \text{H}_2\text{O}$	<code>\ce{H+ + OH- <=> H2O}</code>
	$\text{A} \leftrightarrow \text{A}'$	<code>\ce{\\$A\\$ <-> \\$A'\\$}</code>
opet neželjene	$\text{A} \leftrightarrow \text{A}'$	<code>\ce{\\$A\\$ \leftrightharpoonup \\$A'\\$}</code>
pojave	$\text{A} \leftrightarrow \text{A}'$	<code>\\$ \ce{A} \leftrightharpoonup \ce{A'} \\$</code>
za usporedbu	$\text{A} \longleftrightarrow \text{A}'$	<code>\\$A \longleftrightharpoonup A'\\$</code>
	$\text{CO}_2 + \text{C} \xrightarrow{\alpha} 2\text{CO}$	<code>\ce{CO2 + C ->[\alpha] 2CO}</code>
bez razmaka!	$\text{CO}_2 + \text{C} \longrightarrow [\alpha]2\text{CO}$	<code>\ce{CO2 + C -> [\alpha] 2CO}</code>
	$\text{CO}_2 + \text{C} \xrightarrow[\beta]{\alpha} 2\text{CO}$	<code>\ce{CO2 + C ->[\alpha][\beta] 2CO}</code>
	$\text{CO}_2 + \text{C} \xrightarrow[\text{treba ovako}]{\text{tekst}} 2\text{CO}$	<code>\ce{CO2 + C ->[tekst][\text{treba ovako}] 2CO}</code>
	$\text{A} \xrightarrow{\text{H}_2\text{O}} \text{B}$	<code>\ce{\\$A\\$ ->[\ce{H2O}] \\$B\\$}</code>



JOŠ NEKOLIKO PRIMJERA



$$K = \frac{[\text{Hg}^{2+}][\text{Hg}]}{[\text{Hg}_2^{2+}]}$$

